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Lithiated 3-Tosylpropanal and 4-Tosyl-2-butanone Dimethyl Acetals as β -Acylvinyl Anion Equivalents for the Synthesis of Unsaturated 1,4-Dicarbonyl Compounds and α,β -Butenolides

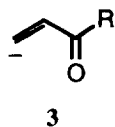
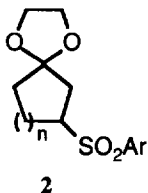
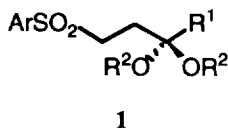
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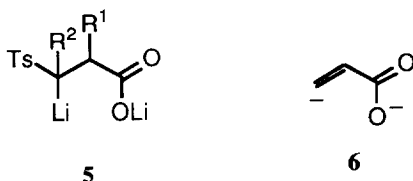
Abstract: The lithiation of 1,1-dimethoxy-3-tosylpropane (**7a**) and 2,2-dimethoxy-4-tosylbutane (**7b**) followed by reaction with acyl chlorides affords, after *p*-toluenesulfonic acid elimination, ene-1,4-dicarbonyl compounds in a stereoselective manner. In the case of compound **7a**, derived from acrolein, sequential monolithiation and reaction with carbonyl compounds give cyclic acetals, which after oxidation and elimination of *p*-toluenesulfonic acid are transformed into α,β -butenolides.

INTRODUCTION

Ketals derived from γ -oxosulfones **1** and **2** are the most simple and stable starting compounds² for the preparation of sulfonyl-stabilized homoenolates. Organolithium compounds derived from protected γ -oxosulfones **1** and **2** are excellent β -acylvinyl anion equivalents **3** and **4** of β -lithiated α,β -unsaturated carbonyl compounds⁵. These types of sulfones are easily prepared by Michael addition of a sulfonic acid to α,β -unsaturated carbonyl compounds followed by protection of the carbonyl group⁶⁻¹⁵. Organolithium derivatives of compounds **1** and **2** have been used in many carbon-carbon bonds forming reactions, e.g.: (a) in alkylation reactions by means of alkyl halides⁶, tosylates⁷, epoxides⁸, and chlorotrimethylsilane⁹ for the synthesis of saturated^{6b} and α,β -unsaturated carbonyl compounds^{6a-e,9} such as (+)-nuciferal^{6a}, *cis*-jasmone^{6e}, 25-hydroxycholesterol^{7a}, (+)-*exo*-brevicomine^{7b} and δ -lactols⁸, (b) in Michael addition reactions to substituted cyclopentenones for the synthesis of 15-oxoprostaglandin F₁¹⁰, (c) in addition to carbonyl compounds to give furans¹¹, such as dendrolasin^{11a}, and 4-hydroxycyclopentenones¹², (d) in the arylation reaction, with a tricarbonylchromium complex, of *N*-silylated indole for the synthesis of clavicipitic acid precursors¹³, and (e) in acylation reaction with esters¹⁴ and ethyl chloroformate¹⁵ for the synthesis of saturated 1,4-dicarbonyl compounds¹⁴ and the γ -hydroxybutenolide moiety of strigol¹⁵, respectively.

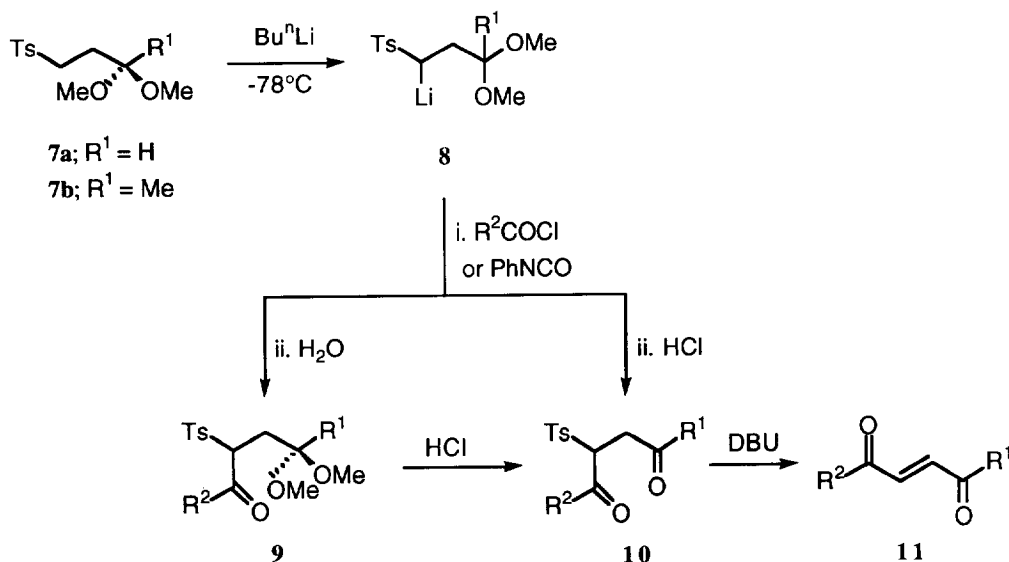


We have recently reported the synthetic applications of lithium 3-lithio-3-tosylalkanoates **5** as β -acylvinyl anion equivalents **6** of β -lithiated α,β -unsaturated carboxylic acids, in the addition to carbonyl compounds and in alkylation and acylation reactions, for the synthesis of α,β -butenolides, α,β and β,γ -unsaturated esters and 4-oxo-2-alkenoates, such as *O*-benzyl derivative of umbelactone, andirolactone, rosefuran lactone and the seco acid of (\pm)-pyrenoforin¹⁶. In connection with these studies, we describe here the synthesis of ene-1,4-dicarbonyl compounds¹⁷ and α,β -butenolides²³ by means of organolithium compounds derived from protected γ -oxosulfones.



RESULTS AND DISCUSSION

When 3-tosylpropanal and 4-tosyl-2-butanone dimethyl acetals (**7a** and **7b**) were allowed to react with *n*-butyllithium at -78°C in THF the corresponding organolithium intermediates **8** were obtained, which reacted with acyl chlorides to give after quenching with water or with 1 N hydrochloric acid monoprotected or unprotected 1,4-dicarbonyl compounds **9** or **10**, respectively (Scheme 1 and Table 1). Starting acetals were prepared⁹ by addition of sodium *p*-toluenesulfonate to acrolein and methyl vinyl ketone followed by acetalization with methyl orthoformate in 85 and 80% yield, respectively. In the case of compound **7a** better yields were obtained when



Scheme 1

Table 1. Preparation of Compounds **9**, **10**, and **12**.

Entry	Starting		Product				
	sulfone	Electrophile	No.	R ¹	R ²	Yield (%) ^a	Mp (°C) ^b or R _f ^c
1	7a	n-PrCOCl	9aa	H	n-Pr	79	0.47
2			10aa			99 ^d	0.45
3	7a	<i>i</i> -PrCOCl	9ab	H	<i>i</i> -Pr	63	0.48
4			10ab			80 ^{d,e}	0.44
5	7a	<i>t</i> -BuCOCl	9ac	H	<i>t</i> -Bu	91	0.46
6			10ac			95	0.34
7	7a	PhCOCl	9ad	H	Ph	75	0.40
8			10ad			73 ^d	0.30
9	7a	PhCH ₂ COCl	9ae	H	PhCH ₂	83 ^e	0.41
10	7a	PhNCO	9af	H	PhNH	81	138-140
11			12a			83 ^d	0.30 ^f
12	7b	n-PrCOCl	10ba	Me	n-Pr	73	77-78
13	7b	<i>i</i> -PrCOCl	10bb	Me	<i>i</i> -Pr	75	118-119
14	7b	<i>t</i> -BuCOCl	10bc	Me	<i>t</i> -Bu	72	117-118
15	7b	PhCOCl	10bd	Me	Ph	88	94-95
16	7b	PhNCO	9bf	Me	PhNH	83	121-123
17			10bf			96 ^d	157-158 ^g
18			12b			36	0.29
19	7b	EtCO ₂ Cl	10bg	Me	EtO	81	0.39

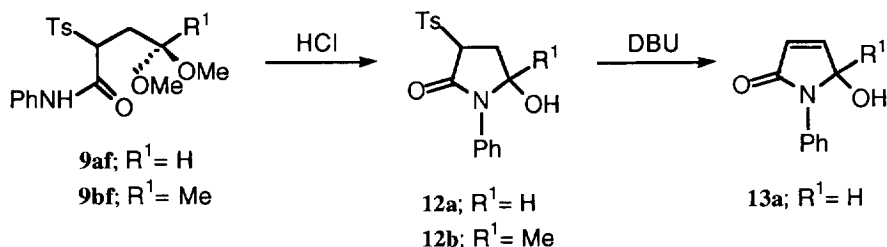
^a Based on starting sulfone **7**, after column chromatography (silica gel, hexane/ether). ^b Hexane/CH₂Cl₂. ^c Hexane/EtOAc: 2/1.

^d Based on pure compound **9**. ^e Hydrolysis was carried out with 50% HClO₄ in dioxane (1/1). ^f Hexane/EtOAc: 1/1. ^g Hexane/EtOAc.

the deprotonation was carried out with two equiv of butyllithium, probably due to the acidity of the hydrogen in α -position to the sulfonyl group in compounds **9**¹⁴, which provokes the protonation of **8a**. However, there is no evidence that a dilithiated intermediate has been formed at -78°C , because after deuterolysis with deuterium oxide or with d^4 -methanol only the monodeuterated acetal derived of **7a** was obtained.

In the case of the 4-oxoaldehydes **10aa-ad**, derived from acrolein, mainly decomposition was observed when they were purified by column chromatography, for this reason were isolated and purified as acetals **9aa-ad** which were transformed into compounds **10** after hydrolysis with 4 N hydrochloric acid in THF (1/1). The methyl vinyl ketone derivatives were isolated and purified as 1,4-diketones **10ba-bg** (Table 1, entries 12-15, 17, and 19).

When phenylisocyanate was used as electrophile with intermediate **8a** followed by hydrolysis with 5% HCl lactam **12a** was isolated as mixture of diastereomers (Scheme 2 and Table 1, entry 11). The same reaction with intermediate **8b** led to the formation of compound **10bf** after acidic hydrolysis, whereas if the reaction was quenched with water compound **9bf** was isolated which after hydrolysis with 5% HCl afforded also lactam **12b** as mixture of diastereomers (Scheme 2 and Table 1, entries 17, 16, and 18, respectively).



Scheme 2

Basic elimination reaction of *p*-toluenesulfonic acid was carried out by treatment of compounds **10** with triethylamine for aldehyde derivatives or 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) for ketone derivatives to give unsaturated 1,4-dicarbonyl compounds **11** (Scheme 1 and Table 2). Lactam **12a** was transformed under the same reaction conditions into the α,β -unsaturated lactam **13a**, whereas the basic elimination of **12b** failed (Scheme 2 and Table 2).

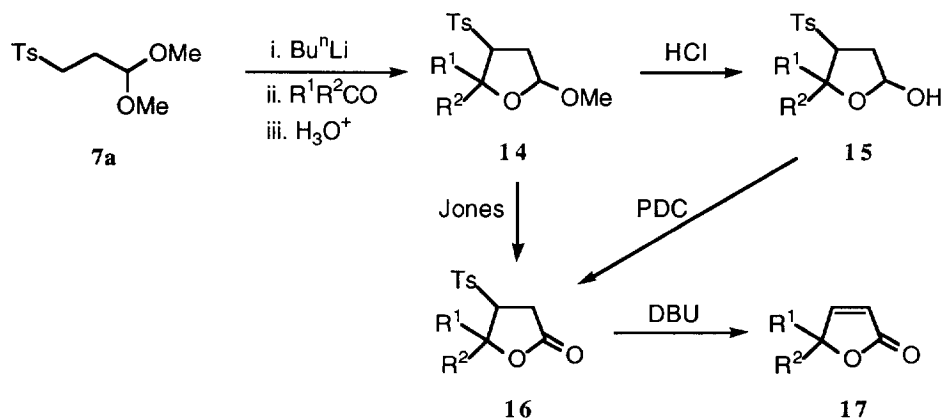
We also have studied the synthesis of α,β -butenolides starting from acrolein derivative **7a**. The reaction of compound **7a** with one equiv of *n*-butyllithium at -78°C followed by addition of carbonyl compounds as electrophiles gave after acidic hydrolysis cyclic acetals **14**, which were treated with Jones reagent to provide lactones **16** in the case of ketones (Scheme 3 and Table 3). When aldehydes were used as electrophiles acetals **14** suffered oxidation to the corresponding 4-oxoacids under Jones oxidation conditions. In these cases compounds **14a** and **14b** have to be first hydrolyzed with 6N HCl to lactols **15**, which were transformed into lactones **16** by oxidation with pyridinium dichromate (PDC)⁴. Butyrolactones **16** were finally treated with DBU to afford α,β -unsaturated butenolides **17**¹⁶ (Scheme 3 and Table 3).

In conclusion, this simple methodology allows the stereoselective synthesis of unsaturated (*E*)-1,4-dicarbonyl compounds and butenolides by means of organolithium compounds derived from γ -oxosulfones and starting from α,β -unsaturated carbonyl compounds.

Table 2. Synthesis of Unsaturated 1,4-Dicarbonyl Compounds **11** and Lactam **13a**.

Entry	No.	R ¹	R ²	Yield (%) ^a	Mp (°C) ^b or <i>R_f</i> ^c	Lit.
1	11aa	H	n-Pr	98 ^d	0.68	24
2	11ac	H	<i>t</i> -Bu	97 ^d	0.61	25
3	11ad	H	Ph	88 ^d	0.49	5b
4	13a	H	PhNH	60	132-135	26
5	11ba	Me	n-Pr	75	0.61	27
6	11bb	Me	<i>i</i> -Pr	96	0.59	22, 28
7	11bc	Me	<i>t</i> -Bu	83	0.63	29
8	11bd	Me	Ph	99	0.56	30
9	11bf	Me	PhNH	68	163-164	31
10	11bg	Me	EtO	62	0.64	32

^a Based on compound **10** or **12** after column chromatography. ^b Hexane/CH₂Cl₂. ^c Hexane/EtOAc: 2/1. ^d The elimination was carried out with Et₃N.



Scheme 3

Table 3. Synthesis of α,β -Butenolides^{16b}.

R ¹	R ²	No.	Yield (%) ^a	No.	Yield (%) ^b	No.	Yield (%) ^c	No.	Yield (%) ^d
<i>i</i> -Pr	H	14a	89	15a	89	16a^e	90 ^f	17a	85
<i>n</i> -C ₅ H ₁₁	H	14b	95	15b	81	16b^g	90 ^f	17b	95
	(CH ₂) ₅	14c	95	-	-	16c	88 ^h	17c	98
Ph	Me	14d	99	-	-	16d^e	82 ^h	17d	99

^a Isolated crude yield, based on starting compound **7a**. ^b Based on compound **14**, after column chromatography. ^c Based on compound **15**. ^d Based on compound **16**, after column chromatography. ^e Diastereomers ratio: 1/1. ^f Isolated crude yield.

^g Diastereomers ratio: 3/2. ^h After column chromatography.

EXPERIMENTAL

General. Melting points were obtained with a Reichert Thermovar apparatus and are uncorrected. IR spectra were obtained as films in a Pye Unicam SP3-200 spectrophotometer. ¹H and ¹³C spectra were recorded on a Bruker AC-300 spectrometer with SiMe₄ as internal standard and using CDCl₃ as solvent. ¹³C-NMR assignments were made on the basis of DEPT experiments. MS spectra were measured in a Hewlett-Packard 5988A (EI, 70eV). Elemental analyses were performed by the Microanalyses Service at the University of Alicante. Chromatographic analysis (GLC) were determined with a Hewlett-Packard HP-5890 instrument equipped with a 25 m WCOT capillary column (0.22 mm diam., 0.2 μ m film thickness OV-101 stationary phase) using nitrogen (2 ml/min) as the carrier gas, T_{injector}=270°C, T_{column}=60°C, and 60-270 (15°C/min). Thin layer chromatography (TLC) was carried out on Schleicher & Schuell F1500/LS 254 plates coated with a 0.2 mm layer of silica gel and UV visualization. Column chromatography was performed using silica gel 60 of 70-270 mesh and hexane/ether as eluant. All starting materials were commercially available (Aldrich, Fluka) of the best grade and were used without further purification. THF was dried with LiAlH₄ under argon atmosphere. Methanol was dried with magnesium turnings under argon.

Preparation of γ -Oxosulfones Acetals **7a and **7b**. General Procedure.** A mixture of sodium *p*-toluenesulfinate (6.0 g, 33 mmol) and the α,β -unsaturated carbonyl compound (30 mmol) in acetic acid (45 ml) was stirred for 12 h. When methyl vinyl ketone was used as carbonyl compound, the reaction mixture was stirred under reflux for 1 d in acetic acid (2.0 ml, 33 mmol) and water (20 ml). Then, ether (150 ml) was added and the reaction mixture was washed with saturated aqueous NaHCO₃ and water. The organic layer was dried over Na₂SO₄, evaporated and the resulting residue was used in the ketalization reaction without further purification. A solution of the γ -oxosulfone (10 mmol), trimethyl orthoformate (2.26 ml, 20 mmol) and *p*-toluenesulfonic acid (80 mg, 0.6 mmol) in dry methanol (35 ml) was stirred 20 h at 50°C. Then, solvents were evaporated *in vacuo*

(15 torr), the resulting residue was dissolved in ether (50 ml) and the organic layer was washed with saturated aqueous NaHCO₃ and water, dried (Na₂SO₄) and evaporated to yield compounds **7a** and **7b**, which were obtained in 85% and 80% yield respectively.

1,1-Dimethoxy-3-tosylpropane (7a): *R*_f 0.37 (Hex/AcOEt 2:1) ν 2820 (CHO), 1300 and 1140 cm⁻¹ (SO₂); δ _H 1.98 (m, 2H, CH₂CH), 2.44 (s, 3H, CH₃Ar), 3.15 (m, 2H, CH₂S), 3.28 (s, 6H, 2xCH₃O), 4.40 (t, *J*=5.0Hz, 1H, CH₂CH), 7.36 and 7.77 (2d, *J*=8.0Hz, 4H, ArH); δ _C 21.3 (CH₃Ar), 26.0 (CH₂CH), 51.4 (CH₂S), 53.4 (2xCH₃O), 102.3 (CHCH₂), 127.8, 129.7, 135.8 and 144.5 (ArC); *m/z* 258 (*M*⁺, <1%), 91 (19), 75 (100), 71 (51), 65 (13) and 41 (13) (Found: *M*⁺ 258.0923. Calcd. for C₁₂H₁₈O₄S, 258.0926).

2,2-Dimethoxy-4-tosylbutane (7b): *R*_f 0.40 (Hex/AcOEt 2:1) ν 2820 (CHO), 1290 and 1145 cm⁻¹ (SO₂); δ _H 1.20 (s, 3H, CH₃CCH₂), 1.98 (m, 2H, CH₂CCH₃), 2.45 (s, 3H, CH₃Ar); 3.07-3.48 (m with s at 3.11, 8H, CH₂S and 2xCH₃O), 7.36 and 7.79 (2d, *J*=8.0Hz, 4H, ArH); δ _C 21.0, 21.6 (CH₃Ar and CH₃CCH₂), 29.5 (CH₂CCH₃), 48.2 (2xCH₃O), 52.3 (CH₂S), 100.2 (CCH₂), 128.1, 129.9, 136.0 and 144.9 (ArC); *m/z* 241 (*M*⁺-OCH₃, 10%), 101 (20), 91 (51), 89 (100), 86 (13), 85 (80), 71 (14), 65 (29), 55 (26) and 43 (62) (Found: *M*⁺-OCH₃ 241.0903. Calcd. for C₁₂H₁₇O₃S, 241.0898).

Preparation of Compounds 9, 10 and 12. General Procedure. To a solution of ketal **7a** or **7b** (1 mmol) in dry THF (5 ml) was slowly added a 1.6 M solution of *n*-BuLi (1.25 ml, 2 mmol or 0.63 ml, 1 mmol, respectively) at -78°C under Ar. The resulting yellow solution was stirred for 1h, and then the electrophile was added. The reaction mixture was allowed to rise rt overnight, quenched with water (20 ml) and extracted with ether (3x20 ml). The organic layer was dried (Na₂SO₄) and evaporated to give crude compounds **9**, which were purified by column chromatography on silica gel. In the case of ketal **7b** the reaction was quenched with 2N HCl (5 ml) and stirred 1h at rt., extracted with ether (3x15 ml), dried (Na₂SO₄) and evaporated to give crude compounds **10**, which were purified by column chromatography. Pure compounds **9** (0.5 mmol) were dissolved in THF (5 ml) and aqueous 4N HCl (5 ml) was added and stirred at rt until complete hydrolysis. Then water (20 ml) was added and the resulting solution was extracted with ether (3x20 ml), the ethereal layers were dried over Na₂SO₄ and evaporated to yield pure compounds **10**. Pure compound **9af** or **9bf** (0.5 mmol) was dissolved in THF (5 ml) and aqueous 6N HCl (5 ml) was added and stirred under reflux, extracted with ether (3x20 ml), dried over Na₂SO₄ and evaporated to yield pure compounds **12**, which were purified by column chromatography. Yields and physical data are included in Table 1, analytical and spectral data follow.

1,1-Dimethoxy-3-tosyl-4-heptanone (9aa): ν 2820 (CHO), 1710 (C=O), 1300 and 1140 cm⁻¹ (SO₂); δ _H 0.91 (t, *J*=7.4Hz, 3H, CH₃CH₂), 1.58 (sext, *J*=7.2Hz, 2H, CH₂CH₃), 2.20 (m, 2H, CH₂C=O), 2.45 (s, 3H, CH₃Ar), 2.58, 2.80 (2m, 2H, CH₂CH), 3.19, 3.25 (2s, 6H, 2xCH₃O), 4.28 (m, 2H, CHS and CHCH₂), 7.35 and 7.65 (2d, *J*=8.1Hz, 4H, ArH); δ _C 13.4 (CH₃CH₂), 16.5 (CH₂CH₃), 21.6 (CH₃Ar), 30.6 (CH₂CH), 46.9 (CH₂C=O), 53.2, 54.1 (2xCH₃O), 70.1 (CHS), 102.0 (CHOCH₃), 129.4, 129.6, 133.3, 145.4 (ArC) and 201.7 (C=O); *m/z* 297 (*M*⁺-OCH₃, 1%), 173 (29), 75 (100) and 71 (66) (Found: *M*⁺-OCH₃ 297.1161. Calcd. for C₁₅H₂₁O₄S 297.1161).

2-Methyl-6,6-dimethoxy-4-tosyl-3-hexanone (9ab): ν 2820 (CHO), 1710 (C=O), 1310 and 1140 cm⁻¹ (SO₂); δ _H 1.11, 1.12 [2d, *J*=6.8Hz, 6H, (CH₃)₂CH], 2.16 (m, 2H, CH₂), 2.44 (s, 3H, CH₃Ar), 3.05 (m, 1H, CHC=O), 3.18, 3.23 (2s, 6H, 2xCH₃O), 4.19 (t, *J*=5.0Hz, 1H, CHO), 4.52 (dd, *J*=10.1, 3.4Hz, 1H, CHS), 7.34 and 7.64 (2d, *J*=8.1Hz, 4H, ArH); δ _C 17.1, 18.0 [(CH₃)₂CH], 21.4 (CH₃Ar), 30.5 (CH₂), 42.5

(CHC=O), 52.6, 53.7 (2xCH₃O), 68.4 (CHS), 101.4 (CHOCH₃), 129.3, 129.5, 133.2, 145.2 (ArC) and 205.4 (C=O); *m/z* 297 (*M*⁺-OCH₃, 1%), 173 (32), 141 (14), 129 (40), 109 (12), 99 (10), 91 (30), 81 (14), 75 (100), 71 (60), 65 (15), 55 (10), 43 (26) and 41 (16).

6,6-Dimethoxy-2,2-dimethyl-4-tosyl-3-hexanone (9ac): ν (CHCl₃) 2850 (CHO), 1695 (C=O), 1310 and 1140 cm⁻¹ (SO₂); δ_{H} 1.25 [s, 9H, (CH₃)₃C], 2.08 (ddd, *J*=13.7, 9.1, 5.3Hz, 1H, 1xCH₂), 2.15 (ddd, *J*=13.7, 5.3, 4.2Hz, 1H, 1xCH₂), 2.45 (s, 3H, CH₃Ar), 3.19, 3.25 (2s, 6H, 2xCH₃O), 4.24 (t, *J*=5.3Hz, 1H, CHCH₂), 4.84 (dd, *J*=9.1, 4.2Hz, CHS), 7.35 and 7.65 (2d, *J*=8.3Hz, 4H, ArH); δ_{C} 21.6 (CH₃Ar), 27.1 [(CH₃)₃C], 32.9 (CH₂), 45.5 (CCH₃), 52.8, 54.0 (2xCH₃O), 65.0 (CHS), 101.6 (CHOCH₃), 129.5, 129.8, 133.6, 145.2 (ArC) and 208.7 (C=O); *m/z* 299 (*M*⁺-OCH₃, 1%), 175 (30), 143 (18), 85 (60), 75 (100), 57 (81), 42 (26) and 40 (41).

4,4-Dimethoxy-1-phenyl-2-tosyl-1-butanone (9ad): ν (CHCl₃) 2830 (CHO), 1675 (C=O), 1305 and 1145 cm⁻¹ (SO₂); δ_{H} 2.35-2.50 (m with s at 2.40, 5H, CH₃Ar and CH₂), 3.13, 3.22 (2s, 6H, 2xCH₃O), 4.30 (t, *J*=7.0Hz, 1H, CHCH₂), 5.25 (dd, *J*=9.0, 4.5Hz, 1H, CHS) and 7.25-7.95 (m, 9H, ArH); δ_{C} 21.8 (CH₃Ar), 31.7 (CH₂), 53.6, 54.1 (2xCH₃O), 65.4 (CHS), 102.1 (CHOCH₃), 128.2, 128.4, 128.7, 129.4, 129.5, 133.2, 134.0, 145.4 (ArC) and 192.5 (C=O). *m/z* 331 (*M*⁺-OCH₃, <1%), 207 (23), 175 (10), 147 (11), 105 (32), 91 (15), 77 (18), 75 (100) and 71 (13).

5,5-Dimethoxy-1-phenyl-3-tosyl-2-pentanone (9ae): ν (CHCl₃) 1710 (C=O), 1300 and 1135 cm⁻¹ (SO₂); δ_{H} 2.13 (m, 2H, CH₂CH), 2.36 (s, 3H, CH₃Ar), 3.00, 3.13 (2s, 6H, 2xCH₃O), 3.92, 3.99 (2d, *J*=15.2Hz, 2H, CH₂C=O), 4.05 (t, *J*=4.8Hz, 1H, CHCH₂), 4.38 (dd, *J*=10.7, 2.7Hz, 1H, CHS) and 7.00-7.70 (m with 2d at 7.10 and 7.55, *J*=8.2Hz, 9H, ArH); δ_{C} 21.6 (CH₃Ar), 30.7 (CH₂CH), 51.4 (CH₂C=O), 53.6, 53.9 (2xCH₃O), 69.3 (CHS), 101.9 (CHOCH₃), 127.2, 128.5, 129.5, 129.7, 132.9, 133.2, 145.5 (ArC) and 199.2 (C=O); *m/z* 360 (*M*⁺-CH₃OH, <1%), 183 (21), 173 (13), 156 (15), 155 (15), 145 (14), 139 (27), 117 (15), 115 (16), 107 (20), 105 (64), 99 (16), 92 (27), 91 (100), 89 (14), 85 (14), 79 (16), 77 (41), 71 (57), 65 (32), 66 (10), 55 (17), 51 (13) and 41 (12).

N-Phenyl-4,4-dimethoxy-2-tosylbutanamide (9af): ν (CHCl₃) 3310 (NH), 2800 (CHO), 1655 (C=O), 1310 and 1130 cm⁻¹ (SO₂); δ_{H} 2.30 (m, 2H, CH₂), 2.42 (s, 3H, CH₃Ar), 3.29, 3.32 (2s, 6H, 2xCH₃O), 4.10 (dd, *J*=8.8, 4.9Hz, 1H, CHCH₂), 4.50 (dd, *J*=6.3, 5.1Hz, 1H, CHS), 7.15 (tt, *J*=7.4, 1.2Hz, 1H, *p*-PhN), 7.33 (m, 4H, *m*-PhN, 2x *p*-CH₃Ph), 7.72 (d, *J*=8.2Hz, 2H, 2x *p*-CH₃Ph) and 8.25 (broad s, 1H, NH); δ_{C} 21.7 (CH₃Ar), 29.8 (CH₂), 52.6, 54.1 (2xCH₃O), 68.1 (CHS), 101.7 (CHOCH₃), 120.1, 125.0, 129.1, 129.2, 130.0, 132.9, 137.2, 145.8 (ArC) and 161.7 (C=O); *m/z* 377 (*M*⁺, 3%), 281 (18), 236 (12), 190 (33), 158 (20), 139 (11), 132 (20), 129 (36), 101 (18), 93 (59), 92 (19), 91 (53), 89 (21), 83 (12), 77 (19), 75 (100), 71 (19), 69 (13), 65 (31), 57 (15), 55 (33), 47 (25), 45 (11), 43 (11) and 41 (20). Anal. Calcd. for C₁₉H₂₃O₅SN: C, 60.46; H, 6.14. Found: C, 60.46; H, 6.39.

N-Phenyl-4,4-dimethoxy-2-tosylpentanamide (9bf): ν (CHCl₃) 3330 (NH), 1670 (C=O), 1310 and 1140 cm⁻¹ (SO₂); δ_{H} 1.24 (s, 3H, CH₃CCH₂), 2.40 (m with s at 2.42, 5H, CH₃Ar, CH₂), 3.14, 3.19 (2s, 6H, 2xCH₃O), 4.09 (dd, *J*=7.2, 4.7Hz, 1H, CHS), 7.13 (tt, *J*=7.0, 1.3Hz, 1H, *p*-PhN), 7.31 (m, 4H, *m*-PhN, 2x *p*-CH₃Ph), 7.48 (m, 2H, 2xPhN), 7.72 (d, *J*=8.5Hz, 2H, 2x *p*-CH₃Ph) and 8.21 (broad s, 1H, NH); δ_{C} 21.2, 21.7 (CH₃CCH₂ and CH₃Ar), 33.1 (CH₂), 48.5, 48.7 (2xCH₃O), 69.1 (CHS), 100.4 (COCH₃), 120.1, 124.9, 129.0, 129.4, 129.8, 133.0, 137.4, 145.6 (ArC) and 162.4 (C=O); *m/z* 391 (*M*⁺, 1%), 295 (31), 204 (48), 188 (15), 172 (24), 144 (76), 143 (23), 132 (46), 111 (17), 97 (14), 94 (10), 93 (38), 92 (20), 91 (52), 89 (100),

83 (23), 77 (18), 65 (27), 55 (30) and 43 (78). Anal. Calcd. for $C_{20}H_{25}O_5SN$: C, 61.36; H, 6.44. Found: C, 61.32; H 6.51.

4-Oxo-3-tosylheptanal (10aa): ν ($CHCl_3$) 2710 (CHO), 1710 (C=O), 1330 and 1140 cm^{-1} (SO_2); δ_H 0.92 (t, $J=7.4Hz$, 3H, CH_3CH_2), 1.61 (m, 2H, CH_2CH_3), 2.47 (s, 3H, CH_3Ar), 2.85 (m, 2H, $CH_2C=O$), 3.15 (dd, $J=18.8, 2.9Hz$, 1H, $1xCH_2CH$), 3.24 (dd, $J=18.8, 11.0Hz$, 1H, $1xCH_2CH$), 4.58 (dd, $J=11.0, 2.9Hz$, 1H, CHS), 7.38, 7.65 (2d, $J=8.2Hz$, 4H, ArH) and 9.58 (s, 1H, CHO); δ_C 13.3 (CH_3CH_2), 21.7 (CH_3Ar), 41.9 (CH_2CH), 46.6 (CH_2CO), 68.6 (CHS), 129.2, 129.9, 133.1, 145.9 (ArC), 196.7 (CH=O) and 200.8 (C=O); m/z 281 (M^+-1 , 5%), 210 (25), 173 (14), 155 (25), 139 (52), 125 (13), 119 (10), 99 (17), 97 (12), 92 (20), 91 (100), 89 (13), 83 (11), 79 (10), 77 (14), 71 (63), 69 (10), 65 (37), 63 (10), 57 (14), 55 (27), 43 (39) and 41 (19).

5-Methyl-4-oxo-3-tosylhexanal (10ab): ν ($CHCl_3$) 2710 (CHO), 1720, 1700 (C=O), 1300 and 1140 cm^{-1} (SO_2); δ_H 1.13, 1.24 [2d, $J=6.8Hz$, 6H, $(CH_3)_2CH$], 2.20 (m, 1H, $CHC=O$), 2.45 (s, 3H, CH_3Ar), 3.10 (m, 2H, CH_2), 4.84 (d, $J=10.5Hz$, 1H, CHS), 7.37, 7.64 (2d, $J=7.6Hz$, 4H, ArH), and 9.56 (s, 1H, CHO); δ_C 17.4, 18.3 [$(CH_3)_2CH$], 21.6 (CH_3Ar), 41.6 (CH_2), 42.2 ($CHC=O$), 67.0 (CHS), 129.1, 129.9, 133.0, 145.8 (ArC), 196.4 (CH=O) and 204.82 (C=O); m/z 282 (M^+ , 1%), 183 (44), 155 (27), 139 (49), 127 (11), 92 (24), 91 (100), 89 (11), 73 (21), 71 (59), 65 (33), 55 (20), 43 (65) and 41 (21).

5,5-Dimethyl-4-oxo-3-tosylhexanal (10ac): ν 2705 (CHO), 1710, 1695 (C=O), 1300 and 1135 cm^{-1} (SO_2); δ_H 1.33 [s, 9H, $(CH_3)_3C$], 2.47 (s, 3H, CH_3Ar), 3.06 (dd, $J=18.6, 3.9Hz$, 1H, $1xCH_2$), 3.16 (dd, $J=18.6, 9.8Hz$, 1H, $1xCH_2$), 5.15 (dd, $J=9.8, 3.9Hz$, 1H, CHS), 7.37, 7.64 (2d, $J=8.3Hz$, 4H, ArH) and 9.54 (s, 1H, CHO); δ_C 21.7 (CH_3Ar), 27.5 [$(CH_3)_3C$], 44.4 [$(CH_3)_3C$], 45.6 (CH_2), 63.3 (CHS), 129.6, 129.8, 133.2, 145.8 (ArC), 196.5 (CHO) and 208.7 (C=O); m/z 253 ($M^+-C_2H_3O$, 1%), 221 (10), 212 (19), 184 (12), 183 (91), 157 (14), 156 (17), 155 (16), 139 (49), 129 (13), 113 (14), 92 (22), 91 (79), 89 (12), 85 (11), 84 (12), 65 (33), 57 (100), 55 (25), 43 (27) and 41 (54).

4-Phenyl-4-oxo-3-tosylbutanal (10ad): ν ($CHCl_3$) 2710 (CHO), 1710, 1670 (C=O), 1300 and 1140 cm^{-1} (SO_2); δ_H 2.48 (s, 3H, CH_3Ar), 3.48 (m, 2H, CH_2), 5.55 (dd, $J=10.1, 2.9Hz$, 1H, CHS), 7.26-7.99 (m, 9H, ArH) and 9.65 (s, 1H, CHO); δ_C 21.6 (CH_3Ar), 42.2 (CH_2), 67.0 (CHS), 128.5, 129.2, 129.3, 129.7, 133.2, 133.8, 145.7 (ArC), 191.2 (CHO) and 196.7 (C=O); m/z 316 (M^+ , <1%), 133 (28), 105 (100), 92 (12), 91 (36), 77 (48), 65 (18), 55 (26) and 51 (17).

4-Tosyl-2,5-octanodione (10ba): ν (Nujol) 1700 (C=O), 1300 and 1140 cm^{-1} (SO_2); δ_H 0.90 (t, $J=7.3Hz$, 3H, CH_3CH_2) 1.60 (m, 2H, CH_2CH_3), 2.10 (s, 3H, $CH_3C=O$), 2.45 (s, 3H, CH_3Ar), 2.79 (m, 2H, $CH_2C=O$), 3.00 (dd, $J=17.9, 2.6Hz$, 1H, $1xCH_2CH$), 3.23 (dd, $J=17.9, 11.2Hz$, 1H, $1xCH_2CH$), 4.56 (dd, $J=11.2, 2.6Hz$, 1H, CHS), 7.37 and 7.65 (2d, $J=8.6Hz$, 4H, ArH); δ_C 13.13 (CH_3CH_2), 16.37 (CH_2CH_3), 21.43 (CH_3Ar), 29.21 ($CH_3C=O$), 41.31 (CH_2CH), 46.50 ($CH_2C=O$), 69.50 (CHS), 128.94, 129.70, 133.25, 145.52 (ArC), 201.05 and 203.70 ($2xC=O$); m/z 296 (M^+ , <1%), 253 (12), 183 (34), 139 (16), 97 (11), 91 (46), 71 (100), 65 (18), 55 (11) and 43 (71). Anal. Calcd. for $C_{15}H_{20}O_4S$: C, 60.7; H, 6.80. Found: C, 60.93; H 6.61.

6-Methyl-4-tosyl-2,5-heptanedione (10bb): ν (Nujol) 1695 (C=O), 1310 and 1125 cm^{-1} (SO_2); δ_H 1.12, 1.23 [2d, $J=6.9Hz$, 6H, $(CH_3)_2CH$], 2.10 (s, 3H, $CH_3C=O$), 2.46 (s, 3H, CH_3Ar), 2.92 (dd, $J=18.0, 2.7Hz$, 1H, $1xCH_2$), 3.18 (m, 2H, $1xCH_2$ and $CHCH_3$), 4.83 (dd, $J=11.0, 2.7Hz$, 1H, CHS), 7.37 and 7.64 (2d, $J=8.1Hz$, 4H, ArH); δ_C 17.4, 19.1 ($2xCH_3CH$), 21.6 (CH_3Ar), 29.6 ($CH_3C=O$), 41.1 (CH_2), 42.2

(CHC=O), 68.3 (CHS), 129.0, 129.8, 133.4, 145.7 (ArC), 203.5 and 205.3 (2xC=O); m/z 296 (M^+ , <1%), 253 (21), 232 (15), 283 (73), 157 (45), 155 (18), 139 (78), 97 (38), 92 (15), 91 (90), 71 (70), 65 (29), 55 (18), 43 (100) and 41 (16). Anal. Calcd. for $C_{15}H_{20}O_4S$: C, 60.79; H, 6.80. Found: C, 60.70; H 6.94.

6,6-Dimethyl-4-tosyl-2,5-heptanedione (10bc): ν (Nujol) 1710, 1685 (C=O), 1300 and 1125 cm^{-1} (SO_2); δ_H 1.33 [s, 9H, $(CH_3)_3C$], 2.06 (s, 3H, $CH_3C=O$), 2.45 (s, 3H, CH_3Ar), 3.01 (dd, $J=18.1, 3.2Hz$, 1H, 1x CH_2), 3.14 (dd, $J=18.1, 10.3Hz$, 1H, 1x CH_2), 5.12 (dd, $J=10.3, 3.2Hz$, 1H, CHS), 7.36 and 7.64 (2d, $J=8.0Hz$, 4H, ArH); δ_C 21.4 (CH_3Ar), 27.3 (3x CH_3C), 29.1 ($CH_3C=O$), 43.8 (CH_2), 45.3 (CC=O), 64.3 (CHS), 129.2, 129.5, 133.3, 145.3 (ArC), 203.4 and 208.9 (2xC=O); m/z 295 (M^+-CH_3 , <1%), 253 (14), 183 (100), 157 (38), 139 (71), 97 (24), 92 (11), 91 (60), 65 (19), 67 (33), 55 (10), 43 (42) and 41 (19). Anal. Calcd. for $C_{16}H_{22}O_4S$: C, 61.91; H, 7.14. Found: C, 61.74; H 7.36.

1-Phenyl-2-tosyl-1,4-pentanodione (10bd): ν ($CHCl_3$) 1710, 1670 (C=O), 1315 and 1150 cm^{-1} (SO_2); δ_H 2.14 (s, 3H, $CH_3C=O$), 2.37 (s, 3H, CH_3Ar), 3.29 (dd, $J=18.1, 2.9Hz$, 1H, 1x CH_2), 3.50 (dd, $J=18.1, 10.7Hz$, 1H, 1x CH_2), 5.52 (dd, $J=10.7, 2.9Hz$, 1H, CHS) and 7.12-8.15 (m, 9H, ArH); δ_C 21.5 (CH_3Ar), 29.4 ($CH_3C=O$), 41.7 (CH_2), 65.4 (CH), 128.4, 129.0, 129.1, 129.6, 132.9, 133.5, 136.5, 145.5 (ArC) 191.5 and 203.8 (2xC=O); m/z 330 (M^+ , <1%), 266 (15), 223 (14), 158 (11), 155 (11), 133 (25), 106 (16), 105 (100), 91 (46), 77 (95), 65 (21), 55 (14), 51 (12) and 43 (43). Anal. Calcd. for $C_{18}H_{18}O_4S$: C, 65.43; H, 5.49. Found: C, 65.49; H 5.73.

N-Phenyl-4-oxo-2-tosylpentanamide (10bf): ν ($CHCl_3$) 3290 (NH), 1705, 1665 (C=O), 1305 and 1145 cm^{-1} (SO_2); δ_H 2.19 (s, 3H, $CH_3C=O$), 2.43 (s, 3H, CH_3Ar), 2.97 (dd, $J=17.6, 2.8Hz$, 1H, 1x CH_2), 3.24 (dd, $J=17.6, 10Hz$, 1H, 1x CH_2), 4.57 (dd, $J=10.4, 2.8Hz$, 1H, CHS), 7.37 (m, 7H, ArH), 7.70 (d, $J=8.6Hz$, 2H, ArH) and 8.36 (br s, 1H, NH); δ_C 21.7 (CH_3Ar), 29.9 ($CH_3C=O$), 39.8 ($CH_2C=O$), 67.1 (CHS), 120.1, 125.1, 129.0, 129.1, 130.1, 132.9, 137.1, 146.1 (ArC), 160.7 (NC=O) and 203.6 (C=O); m/z 345 (M^+ , 2%), 238 (17), 139 (12), 97 (12), 94 (16), 93 (73), 65 (37), 55 (47), 51 (10), 43 (100) and 41 (10). Anal. Calcd. for $C_{18}H_{19}O_2NS$: C, 62.59; H, 5.54. Found: C, 62.00; H 5.74.

Ethyl 4-Oxo-3-tosylpentanoate (10bg): ν ($CHCl_3$) 1715 (C=O), 1320 and 1145 cm^{-1} (SO_2); δ_H 1.11 (t, $J=7.2Hz$, 3H, CH_3CH_2), 2.22 (s, 3H, $CH_3C=O$), 2.47 (s, 3H, CH_3Ar), 3.23 (dd $J=18.2, 4.1Hz$, 1H, 1x CH_2CH), 3.34 (dd, $J=18.2, 10.1Hz$, 1H, 1x CH_2CH), 4.05 (q, $J=7.2Hz$, 2H, OCH_2), 4.42 (dd, $J=18.2, 10.1Hz$, 1H, CH), 7.37 and 7.75 (2d, $J=7.2Hz$, 4H, ArH); δ_C 13.6 (CH_3CH_2), 21.7 (CH_3Ar), 29.8 ($CH_3C=O$), 39.3 ($CH_2C=O$), 62.3 (CHS), 65.8 (CH_2O), 128.9, 129.7, 134.6, 145.5 (ArC), 165.1 (COO) and 203.6 (C=O); m/z 283 (M^+-CH_3 , <1%), 234 (15), 191 (36), 157 (13), 155 (62), 150 (26), 143 (21), 139 (31), 101 (60), 97 (13), 92 (14), 91 (100), 73 (28), 65 (28), 55 (23) and 43 (84).

Preparation of Unsaturated 1,4-Dicarbonyl Compounds 11 and Lactam 13a. General Procedure.

To a solution of compound **10** or **12a** (0.5 mmol) in dry CH_2Cl_2 (5 ml) at $0^\circ C$ was added DBU (0.082 ml, 0.55 mmol) and stirred at rt until elimination was complete. The reaction mixture was poured into an aqueous saturated solution of $NaHCO_3$ (25 ml) and extracted with ether (3x10 ml). The ethereal layers were washed with 2N aq HCl and brine, dried (Na_2SO_4), and evaporated to yield crude compounds **11** and **13a**, which were purified by column chromatography on silica gel. Yields and physical data are include in Table 2, spectral data follow.

(E)-4-Oxo-2-heptenal (11aa)²³: ν 2860, 2730 (CHO), 1710, 1680 (C=O), 1590 and 980 cm^{-1} (CH=C); δ_H 0.97 (t, $J=7.3Hz$, 3H, CH_3), 1.70 (sextet, $J=7.3Hz$, 2H, CH_2CH_3), 2.69 (t, $J=7.3Hz$, 2H, $CH_2C=O$), 6.78

(dd, $J=16.5$, 6.9Hz , 1H , CHCHO), 6.89 (d, $J=16.5\text{Hz}$, 1H , CHC=O) and 9.78 (d, $J=6.9\text{Hz}$, 1H , CHO); δ_{C} 13.6 (CH_3), 17.1 (CH_2CH_3), 43.0 ($\text{CH}_2\text{C=O}$), 137.3 ($2\times\text{CH}$), 193.4 (CHO) and 200.0 (C=O); m/z 98 (M^+ - C_2H_4 , 16%), 97 (47), 84 (22), 83 (42), 56 (15), 55 (100), 54 (14), 53 (11), 43 (93), 42 (27) and 41 (99).

(E)-5,5-Dimethyl-4-oxo-2-hexenal (11ac)²⁴: ν 3020 , 1615 , 980 (CH=C), 2730 (CHO) and 1690 cm^{-1} (C=O); δ_{H} 1.24 [s, 9H , (CH_3) $_3\text{C}$], 6.91 (dd, $J=15.5$, 7.6Hz , 1H , CHCHO), 7.38 (d, $J=15.5\text{Hz}$, 1H , CHC=O) and 9.79 (d, $J=7.6\text{Hz}$, 1H , CHO); δ_{C} 25.7 [(CH_3) $_3\text{C}$], 43.7 [(CH_3) $_3\text{C}$], 138.3 , 140.7 (CH=CH), 192.8 and 203.5 (C=O); m/z 140 (M^+ , $<1\%$), 84 (100), 57 (71), 55 (17) and 41 (47).

(E)-4-Oxo-4-phenyl-2-butenal (11ad)²⁵: ν (CHCl_3) 3020 , 1615 , 980 (CH=C), 2730 (CHO) and 1690 cm^{-1} (C=O); δ_{H} 6.99 (dd, $J=15.8$, 7.4Hz , 1H , CHCHO), 7.54 (tt, $J=7.6$, 1.3Hz , 2H , $m\text{-Ph}$), 7.66 (m, 1H , $p\text{-Ph}$), 7.71 (d, $J=15.8$, 1H , CHC=O), 8.93 (m, 2H , $o\text{-Ph}$) and 9.89 (d, $J=7.4\text{Hz}$, 1H , CHO); δ_{C} 128.9 , 129.0 , 134.1 , 136.4 , 139.1 , 142.0 (ArC and CH=CH), 191.1 and 192.7 (C=O); m/z 160 (M^+ , 32%), 131 (10), 105 (91), 77 (100), 74 (13), 54 (10), 53 (10), 51 (68) and 50 (33).

(E)-3-Octen-2,5-dione (11ba)²⁶: ν (CHCl_3) 3025 , 1590 , 980 (CH=C) and 1665 cm^{-1} (C=O); δ_{H} 0.96 (t, $J=7.4\text{Hz}$, 3H , CH_2CH_3), 1.18 (m, 2H , CH_2CH_3), 2.37 (s, 3H , $\text{CH}_3\text{C=O}$), 2.64 (t, $J=7.2\text{Hz}$, 2H , $\text{CH}_2\text{C=O}$), 6.81 and 6.85 (2 deformed d, $J=13.6\text{Hz}$, 2H , CH=CH); δ_{C} 13.6 (CH_3CH_2), 17.2 (CH_2CH_3), 28.1 ($\text{CH}_3\text{C=O}$), 43.18 ($\text{CH}_2\text{C=O}$), 136.8 , 137.2 (CH=CH), 198.5 and 200.6 ($2\times\text{C=O}$); m/z 140 (M^+ , $<1\%$), 97 (65), 69 (19), 55 (19), 43 (100) and 41 (22).

(E)-6-Methyl-3-hepten-2,5-dione (11bb)²²: ν (CHCl_3) 3010 , 1615 , 975 (CH=C), 1690 and 1670 cm^{-1} (C=O); δ_{H} 1.16 [d, $J=6.9\text{Hz}$, 6H , (CH_3) $_2\text{CH}$], 2.37 (s, 3H , $\text{CH}_3\text{C=O}$), 2.90 (m, 1H , CHCH_3), 6.89 and 6.95 (2 deformed d, $J=15.9\text{Hz}$, 2H , CH=CH); δ_{C} 19.7 [(CH_3) $_2\text{CH}$], 28.5 ($\text{CH}_3\text{C=O}$), 39.8 (CHCH_3), 135.7 , 136.9 (CH=CH), 198.3 and 203.8 ($2\times\text{C=O}$); m/z 140 (M^+ , 3%), 98 (86), 97 (58), 84 (16), 70 (22), 69 (27), 55 (83), 54 (11), 49 (19), 43 (100), 42 (10) and 41 (31).

(E)-6,6-Dimethyl-3-hepten-2,5-dione (11bc)²⁹: ν 3010 , 1605 , 975 (CH=C) and 1665 cm^{-1} (C=O); δ_{H} 1.20 [s, 9H , (CH_3) $_3\text{C}$], 2.37 (s, 3H , $\text{CH}_3\text{C=O}$), 6.99 and 7.32 (2d, $J=15.5\text{Hz}$, CH=CH); δ_{C} 21.7 [(CH_3) $_3\text{C}$], 29.1 ($\text{CH}_3\text{C=O}$), 43.6 (CC=O), 132.6 , 137.1 (CH=CH), 197.8 and 204.2 ($2\times\text{C=O}$); m/z 154 (M^+ , $<1\%$), 98 (100), 98 (10), 82 (15), 70 (25), 69 (11), 66 (71), 55 (38), 54 (25), 43 (77), 42 (20) and 41 (77).

(E)-1-Phenyl-2-penten-1,4-dione (11bd)³⁰: ν (CHCl_3) 3070 , 1630 , 1600 , 1585 , 980 (CH=C) and 1660 cm^{-1} (C=O); δ_{H} 2.44 (s, 3H , CH_3), 7.09 , 7.70 (2d, $J=15.5\text{Hz}$, 2H , CH=CHC=O), 7.52 (m, 2H , $m\text{-Ph}$), 7.64 (m, 1H , $p\text{-Ph}$) and 7.99 (m, 2H , $o\text{-Ph}$); δ_{C} 29.0 (CH_3), 128.8 , 128.9 , 133.9 , 134.0 , 136.7 , 138.4 (CH=CH and ArC), 190.3 and 197.8 (C=O); m/z 174 (M^+ , 11%), 159 (14), 131 (21), 105 (100), 103 (10), 77 (85), 51 (44), 50 (16) and 43 (43).

N-Phenyl-4-oxo-2-pentenamide (11bf)³¹: ν (KBr) 3320 (NH), 1660 , 1545 (C=O), 1625 , 1600 and 975 cm^{-1} (CH=C); δ_{H} 2.39 (s, 3H , CH_3), 6.99 , 7.21 (2d, $J=15.4\text{Hz}$, 2H , CH=CHC=O), 7.19 (m, 1H , $1\times\text{ArH}$), 7.35 (t, $J=7.5\text{Hz}$, 2H , $2\times\text{ArH}$), 7.63 (d, $J=7.5\text{Hz}$, 2H , $2\times\text{ArH}$) and 8.20 (br s, 1H , NH); δ_{C} 29.6 (CH_3), 120.1 , 125.1 , 129.1 , 134.5 , 138.9 , 137.5 (CH=CH and ArC), 162.6 (NC=O) and 198.0 (C=O); m/z 198 (M^+ , 14), 146 (60), 97 (23), 93 (100), 77 (12), 69 (18), 65 (23) and 43 (74).

Ethyl (E)-4-Oxo-2-pentenoate (11bg)³²: ν (CHCl_3) 3020 , 1635 , 975 (HC=C), 1715 and 1690 cm^{-1} (C=O); δ_{H} 1.33 (t, $J=7.1\text{Hz}$, 1H , CH_3CH_2), 2.37 (s, 3H , $\text{CH}_3\text{C=O}$), 4.28 (q, $J=7.1\text{Hz}$, CH_2), 6.65 and 7.02 (2d, $J=16.1\text{Hz}$, 2H , HC=CH); δ_{C} 14.01 (CH_3), 27.97 ($\text{CH}_3\text{C=O}$), 61.35 (CH_2), 131.51 , 139.84 (HC=CH),

165.36 (COO) and 197.53 (C=O); m/z 142 (M^+ , 13%), 127 (100), 124 (17), 99 (39), 97 (65), 71 (12), 89 (14), 55 (21) and 43 (38).

N-Phenyl-5-hydroxy-2-pyrrolinone (13a)^{5b}: ν (CHCl₃) 3180 (OH, NH), 1680 (C=O) and 1590 cm⁻¹ (O=CNH); δ_H 3.08 (br s, 1H, OH), 5.98 (br s, 1H, CHOH), 6.20 (d, $J=6.0$ Hz, 1H, CHC=O), 7.03 (dd, $J=6.0$, 1.7Hz, 1H, CHCHOH), 7.19 (t, $J=7.8$ Hz, 1H, *p*-Ph), 7.39 (t, $J=7.8$ Hz, 2H, *m*-Ph) and 7.64 (d, $J=7.8$ Hz, 2H, *o*-Ph); δ_C 84.1 (CHOH), 113.2, 121.2, 125.2, 129.2, 136.6, 144.9 (HC=CH and ArC) and 168.4 (C=O); m/z 176 (M^++1 , 13%), 175 (M^+ , 100), 146 (43), 130 (44), 128 (16), 119 (12), 118 (13), 104 (13), 93 (61), 91 (10), 77 (57), 55 (26) and 51 (15).

Preparation of Cyclic Acetals 14 and Lactols 15. General Procedure. To a solution of acetal **7a** (1 mmol) in dry THF (5 ml) was added a 1.6 M hexane solution of *n*-BuLi (0.75 ml, 1.2 mmol) at -78°C under Ar. The resulting yellow solution was stirred for 1h at the same temperature and the carbonyl compound was added (1.1 mmol). The reaction mixture was stirred overnight and was quenched with 2N aq HCl (20 ml) and extracted with ether (3x20 ml), washed with water, dried (Na₂SO₄) and solvents were removed under vacuum (15 torr) to yield crude products **14** as diastomeric mixture which were used in the next step without further purification. To a solution of crude compound **14** (1 mmol) in THF (10 ml) was added aqueous 4N HCl (10 ml) and the resulting mixture was stirred at rt. The reaction was extracted with ether (3x20 ml), the organic layer was washed with water, dried (Na₂SO₄) and evaporated to yield crude products **15** as diastomeric mixture which were purified by column chromatography on silica gel.

Preparation of Lactones 16. General Procedure. To a solution of compound **15a,b** (1 mmol) in CH₂Cl₂ (10 ml) at 0°C, PDC (0.56 g, 1.5 mmol) and 4Å molecular sieves were added and the reaction mixture was stirred at rt. When the reaction was complete (TLC), the reaction mixture was purified by column chromatography on silica gel to yield pure compounds **16a,b**. To a solution of compound **15c,d** (1 mmol) in acetone (5 ml) at 0°C, CrO₃ (0.67 g, 6.7 mmol) dissolved in concd. sulfuric acid (1ml) and water (5 ml) were added. The reaction mixture was stirred at rt until complete oxidation. The solvents were removed *in vacuo*, water (20 ml) was added and extracted with ether (3x20 ml). The ethereal layer was washed with brine (20 ml), dried (Na₂SO₄) and evaporated to give γ -lactones **16c,d** as a diastomeric mixture.

Preparation of α,β -Butenolides 17. General Procedure. See general procedure for the preparation of compounds **11** and **13a**. These compounds have been described in reference 16b.

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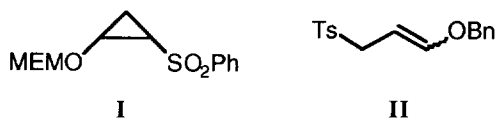
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17. Enedicycarbonyl compounds are versatile synthetic building blocks capable of many subsequent useful chemical manipulations, e. g. (a) dienophiles in Diels-Alder reactions¹⁸, (b) Michael acceptors¹⁹, (c) precursors of 4-methoxycyclopentenones²⁰, and (d) precursors of heterocyclic compounds such as pyrazoles^{18b}. Moreover, they are important features of many natural products having interesting biological

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